ERRATA

'Crystallinity and fusion of low molecular weight α,ω-alkoxy-poly(ethylene oxide): octadecoxy to triacontanoxy end-groups' R. C. Domszy, R. H. Mobbs, Y. -K. Leung, F. Heatley and C. Booth *Polymer* 1979, **20**, 1204–1214

Abstract, line 7, read:
The melting points of the oxyethylene *lamellae* increase markedly as the methylene chain length is increased.

page 1204, 2nd column, line 21, *read*:
analysed by the method of Pickett *et al.*[3]

page 1205, 1st column, line 4: on Melinex polyester film,

1st column, under *Raman scattering*, lines 4–6:
Samples were crystallized at the temperatures indicated in *Table 3*, and, to reduce fluorescent background, were exposed to radiation for 1 h before recording the spectra.

page 1207, 1st column, line 9, *read*:
that the methylene chains crystallize, and serve to locate . . .

page 1207, 2nd column, line 2, *read*:
. . . on the basis of the preceding results of . . .

page 1207, *Table 4* should appear as:

Table 4 Melting points (T_m) determined by dilatometry and d.s.c.

	Dilatometry			D.s.c.	
	T_c = 25°C	35°C	45°C	25°C	45
18–45–18	50.4	50.4	50.6	50.0	–
21–45–21	53.5	53.5	53.6	53.5	53.5
26–45–26	35.0, 56.3	56.4	55.3, 58.0	34.5, 55.5	58.0
30–45–30	49.5, 66.7	–	49.5, 66.8	49.5, 66.0	49.5, 66.0

page 1208, 2nd column, line 28:
. . . may well be a result of the very . . .
The subheading should *read*: *Nuclear magnetic resonance*

page 1209, 1st column, lines 20–21 *read*:
We find $X_{oe} \simeq 0.85$ when the oxyethylene block crystallizes alone in extended-chain conformation ($n \leqslant 4$).

page 1209, 2nd column, under *Variation of structure with crystallization temperature* line 5, *read*:
Provided that microphase separation *in the melt* is not . . .
under *Melting points*, line 4, *read*: extended-chain to folded-chain

page 1209–1210, for ≅ read ≃ throughout
Figure 4 upper line is (a) and lower line is (b)

page 1210, 2nd column, line 15, *read*: *segregation*
References

6 *Polymer* 1979, **20**, 778
14 Beaumond, R. H., Clegg, B., Gee, G., Herbert, J. B. M., Marks, D. J., Roberts, R. C. and Sims, D. *Polymer* 1966, 7, 401
23 Schneider, A. K. and Spielman, M. A. *J. Biol. Chem.* 1942, **142**, 345
37 Swann Jr., S., . . .
41 Brensch, F. L., Baykut, F. and Ozeris, S. *Fette Seifen Anstrichmittel* 1959, **61**, 891
42 Lukes, R. and Cerný, M. *Collect. Czech. Chem. Commun.* 1958, **23**, 497

page 1211, Appendix A, throughout *read*:
1-bromohexacosane *not l*-bromohexacosane
1-bromoheneicosane (I) *not l*-bromoheneicosane

page 1211, under Appendix A, line 22, *read*: pass it in heptane solution . . .

page 1212, column 1, line 1, *read*: Hopkin and Williams,

page 1212 under *Methyl hydrogen decanedioate (II)*, line 6, *read* distillation gave . . .

page 1212, 2nd column, lines 11 and 12, *read*:
The filtrate was added dropwise (5 min) to the prepared *zinc* chloride solution,
under *Hentriacontanoic acid (IV)*, line 5, *read*:
Water and excess *hydrazine* were then distilled off . . .

page 1213, column 1, line 15, *read*: by Raal *et al.* . . .

column 1, under *1-Bromotriacontane (VII)*, line 31, *read*:
. . . (⩾C–Br) . . .

page 1214, column 1, line 6, *read*: . . . (⩾C–Br).

page 1214 under Appendix B, second equation, *read*:

$$RO^- + R''CH_2CH_2Br \rightarrow ROH + R''CH{=}CH_2 + Br^-$$

We apologize for these errors